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## S-Alkyl Dithioformates as 1,3-Dipolarophiles. Generation of C(2)-Unsubstituted Penems

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## **ABSTRACT**

$$S_{\text{Me}} = S_{\text{MeS}} + S_{\text{H}} = S_{\text{CO}_{2}R} + S_{\text{CO}_{2}R}$$

S-Alkyl dithioformates, generated by a cycloreversion process, react as 1,3-dipolarophiles with  $\beta$ -lactam-based azomethine ylids to provide, after (net) elimination of MeSH, C(2)-unsubstituted penems. The overall cycloreversion/cycloaddition sequence was accelerated by microwave irradiation.

Penems represent a continuing area of interest within the  $\beta$ -lactam area. In particular, 6-exoalkylidene variants such as  $\mathbf{1}^{2a}$  and  $\mathbf{2}^{2b}$  have attracted attention because of their potent activity against class A and class C  $\beta$ -lactamases and bacterial signal peptidase, respectively, profiles that makes this group of C(2)-unsubstituted penems attractive both for clinical application and as biological probes for  $\beta$ -lactamase structure and function.

We have previously described a versatile entry to bicyclic  $\beta$ -lactams, including C(2)-substituted penams and penems

resulting from the generation of  $\beta$ -lactam-based azomethine ylid reactivity.<sup>4</sup>

Oxazolidinone 3 (PNB = 4-nitrobenzyl) reacts (via a sequential ring cleavage to give azomethine ylid 4 and *then* cycloaddition followed by decarboxylation)<sup>4b,d</sup> with thioketones to provide racemic penams 5 ( $R_1 = R_2 = alkyl$ , aryl). Use of dithiocarboxylates and trithiocarbonates as 1,3-dipolarophiles leads, after net loss of MeSH, to C(2)-

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<sup>(3)</sup> For an example of the use of a novel 6-exoalkylidene variant related to **1** as a probe for the  $\beta$ -lactamase structure and mechanism of action using X-ray crystallography, see: Nukaga, M.; Abe, T.; Venkatesan, A. M.; Mansour, T. S.; Bonomo, R. A.; Knox, J. R. *Biochemistry* **2003**, *42*, 13152–13159

Scheme 1. Azomethine Ylid Strategy for the Synthesis of Penams and Penems

$$\begin{array}{c} \overset{H}{\overset{}{\stackrel{}{=}}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{\overset{}{=}} O \\ \overset{\bullet}{\overset{}{=}} O$$

substituted penems  $\mathbf{6}$  ( $R_1$  = alkyl, aryl, S-alkyl) (Scheme 1).<sup>5</sup> The extension of the azomethine ylid strategy to the synthesis of the C(2)-unsubstituted penem moiety (i.e.,  $\mathbf{6}$ ,  $R_1$  = H) that is associated with  $\mathbf{1}$  and  $\mathbf{2}$  is the focus of this paper.

Achieving this objective required access to dithioformates **7** and an evaluation of the ability of these units to function as effective 1,3-dipolarophiles. Only a very limited range of dithioformates have been reported to date,<sup>6,7</sup> and various approaches to *S*-alkyl dithioformates **7** were evaluated (Scheme 2).

**Scheme 2.** Generation and Trapping of Dithioformates

$$CS_{2} \xrightarrow{\text{i LiBEt}_{3}\text{H}} MeS \xrightarrow{\text{Freagent}} M$$

S-Methyl dithioformate 7a is available by reduction of  $CS_2$  with LiBEt<sub>3</sub>H followed by S-methylation. We found it most convenient (see below) to trap 7a with cyclopentadiene to give the corresponding cycloadduct 8a as a 1.5:1 mixture of exo and endo isomers in 35% yield. The S-benzyl variants

(6) Block, E.; Aslam, M. Tetrahedron Lett. 1985, 26, 2259-2262.

**Scheme 3.** Dithioformates as 1,3-Dipolarophiles<sup>a</sup>

<sup>a</sup> Thermal vs microwave conditions studied: (a) (thermal) **8a**, MeCN, 2 days, 80 °C (19%); (b) (microwave) screw cap pressure vessel, **8a**, PhMe, 55 W, 1 h (76%) or **8a**, emimPF<sub>6</sub> (10 mol %), 55 W, PhMe, 1 h (56%); (c) (microwave) *open* vessel, **8a**, PhMe, 200 W, 5 h (62%) or **8a**, emimPF<sub>6</sub> (10 mol %), 55 W, PhMe, 4 h (40%); (d) (microwave) open vessel, **8b**, PhMe, 200 W, 4 h, (45%); (e) (microwave) screw cap pressure vessel, **8c**, PhMe, emimPF<sub>6</sub> (10 mol %), 55 W, 1 h, (45%).

**7b** and **7c** were best obtained by direct thionation of the corresponding thioformates  $9^9$  and  $10^9$  using Lawesson's reagent. <sup>10</sup> In both cases, the target dithioformates **7b** and **7c** were not isolated but were trapped in situ with cyclopentadiene to give cycloadducts **8b** and **8c**, respectively, in moderate yields for this two-step sequence. <sup>11,12</sup>

Cycloadducts 8a-c were especially attractive for our purposes, representing a potentially controlled supply of the requisite dithioformate (via 4+2 cycloreversion); the retro Diels-Alder reaction provides an in situ source of dipolarophile that is compatible with release of the key azomethine ylid intermediate 4 from oxazolidinone 3.

This strategy was validated, and thermolysis of **8a** in the presence of oxazolidinone **3** provided the racemic cycloadduct **11a** as a 2.5:1 mixture of *exo* and *endo* isomers (Scheme 3). The structure of *exo*-**11a** was confirmed by X-ray crystallography (see Supporting Information).

However, this thermal process (conditions a) did require 2 days to go to completion and this only achieved a very

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<sup>(5) (</sup>a) Planchenault, D.; Wisedale, R.; Gallagher, T.; Hales, N. J. *J. Org. Chem.* **1997**, *62*, 3438. (b) Highly reactive thioaldehydes have also been generated and trapped in situ: Brown, G. A.; Anderson, K. M.; Large, J. M.; Planchenault, D.; Urban, D.; Hales, N. J.; Gallagher, T. *J. Chem. Soc., Perkin Trans. 1* **2001**, 1897–1900.

<sup>(7) (</sup>a) Seyferth, D.; Womack, G. B. *Organometallics* **1984**, *3*, 1891–1897. (b) Gandhi, T.; Nethaji, M.; Jagirdar, B. R. *Inorg. Chem.* **2003**, *42*, 4798–4800. Jagirdar et al. <sup>7b</sup> were able to isolate **7a** by distillation. Block<sup>6</sup> and Seyferth<sup>7a</sup> did not isolate this volatile component but trapped it in situ as a Diels Alder cycloadduct and as an Fe-based coordination complex, respectively.

<sup>(8)</sup> Similar *exo* and *endo* cycloadducts based on thioaldehydes have been characterized previously. Kirby, G. W.; Lochead, A. W. *J. Chem. Soc.*, *Chem. Commun.* **1983**, 1325–1327. Vedejs, E.; Stults, J. S.; Wilde, R. G. *J. Am. Chem. Soc.* **1988**, *110*, 5452–5460.

<sup>(9)</sup> Bax, P. C.; Holsboer, D. H.; Van der Veek, A. P. M. *Recl. Trav. Chim. Pays-Bas* **1971**, *90*, 562–567.

<sup>(10)</sup> S-Benzyl dithioformate (PhCH<sub>2</sub>SC(S)H) could not be prepared using the reduction/S-alkylation strategy associated with 7a but was obtained by direct thionation of the corresponding thioformate (PhCH<sub>2</sub>SC(O)H) using Lawesson's reagent. NMR (CDCl<sub>3</sub>) for PhCH<sub>2</sub>SC(S)H:  $\delta_{\rm H}$  11.28;  $\delta_{\rm C}$  216.8). However, we were unable to isolate this product in an efficient manner, but this served as a model for the preparation of 7b and 7c.

<sup>(11)</sup> **8b** and **8c** were obtained as a 0.6:1 and 1.7:1 mixture of *exo* and *endo* isomers, respectively.

<sup>(12)</sup> One of the issues contributing to the yields obtained for **8b/c** was purification of the desired product from the residues associated with the thionation step. Ley (Ley, S. V.; Leach, A. G.; Storer, R. I. *J. Chem. Soc., Perkin Trans. I* **2001**, 358–361) has reported a solid-phase variant of Lawesson's reagent. In our hands, this reagent worked well for the thionation of amides, but we were unsuccessful in our attempts to thionate *S*-benzyl thioformate (PhCH<sub>2</sub>SC(O)H); see ref 10.

<sup>(13)</sup> Mechanistic studies provide evidence that oxazolidinone 3 is in equilibrium with the carboxylated azomethine ylid 4. This pathway provides an equilibrium concentration of 4, and we have exploited this to trap highly reactive and short-lived 1,3-dipolarophiles.<sup>56</sup>

**Scheme 4.** Generation of a C(2)-Unsubstituted Penem

modest 19% yield of cycloadduct 11a.14 This hurdle was overcome by carrying out the fragmentation of 8a and subsequent 1,3-dipolar cycloaddition step with microwave irradiation (conditions b and c). Using toluene as a solvent, a 76% yield of the target cycloadduct 11a was isolated following irradiation of 3 and 8a.15 Cycloadduct 11a was obtained in 56% yield when the same reaction was carried out for 1 h using 10 mol % of an ionic liquid (emimPF<sub>6</sub>) as an additive. 16 This reaction was carried out using both sealed and open vessel conditions. On the basis of other observations, 14 microwave irradiation is expected to accelerate the retro Diels-Alder reaction of **8a** but may also promote the 1,3-dipolar cycloaddition step between 4 and 7a. Significantly, we have observed considerable rate and yield increases when 3 has been reacted with stable dipolar ophiles, e.g., N-phenyl maleimide, under the same microwave conditions.

A similar cycloreversion/1,3-dipolar cycloaddition sequence was also achieved using cyclopentadiene adducts **8b** and **8c** to provide the bicyclic  $\beta$ -lactams **11b** and **11c** (Scheme 3).

Elaboration of 11a to the corresponding C(2)-unsubstituted penem was carried out by S-oxidation followed by base treatment to give the C(2)-unsubstituted penem 12 in 70% yield (Scheme 4). The base-mediated elimination step was also examined under microwave radiation conditions but provided 12 in a lowered yield (40%).

We were interested in extending the processes outlined in Schemes 3 and 4 to a C(6)-substituted penem. In the event,

**Scheme 5.** Extension to a C(6)-Substituted Variant<sup>a</sup>

<sup>a</sup> Microwave conditions employed: (a) screw cap pressure vessel, **8c**, PhMe, 55 W, 1 h, 25%; (b) open vessel, **8c**, PhMe, 200 W, 4 h, 25%.

the C(6)-O-silylated hydroxyethyl  $\beta$ -lactam-based oxazolidinone **13**<sup>17</sup> reacted with **8c** to give the corresponding dipolar cycloadduct **14c** in 25% yield. Subsequent oxidation and elimination proceeded to give the C(2)-unsubstituted penem **15** in 40% yield (Scheme 5).<sup>18</sup>

In summary, simple S-alkyl dithioformates 7a-c are viable 1,3-dipolarophiles, which can be released in situ, reacting with  $\beta$ -lactam-based oxazolidinones 3 and 13 to provide, after oxidation and elimination, C(2)-unsubstituted penems, represented by 12 and 15. Importantly, the overall cycloreversion/dipolar cycloaddition sequence (e.g.,  $3+8\to11$ , Scheme 3) was accelerated very significantly by microwave irradiation.

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**Supporting Information Available:** Experimental and characterization data for all new compounds and crystallographic details for *exo-11a*. This material is available free of charge via the Internet at http://pubs.acs.org.

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<sup>(14)</sup> We tried to trap **7a** (generated by reduction of CS<sub>2</sub> and S-methylation as outlined in Scheme 2) directly using oxazolidinone **3** and thereby avoiding the need to prepare **8a**. Crude **7a** (in THF) was added to oxazolidinone **3** (MeCN, 80 °C) and cycloadduct **11a** was isolated in 15% yield. However, this yield was obtained after a reaction time of only 1 h, indicating the high inherent reactivity of **7a** as a 1,3-dipolarophile. This observation suggests that cycloreversion of **8** is rate limiting.

<sup>(15)</sup> We used the CEM Discover system as the microwave reactor. Reactions were carried out on 0.25 mmol scale in a screw cap pressure vessel in a PhMe solution in two stages. Stage 1: 150 W, 5 min, max temp 150 °C. Stage 2: 55 W, 60 min, max temp 200 °C. Attempts to accelerate the direct reaction of **7a** with **3** (see ref 14) under microwave conditions led only to decomposition. For recent disclosures of microwave-assisted 1,3-dipolar cycloadditions, see: Cheng, Q.; Zhang, W.; Tagami, Y.; Oritani, T. *J. Chem. Soc., Perkin Trans. I* **2001**, 452–456. Wilson, N. S.; Sarko, C. R.; Roth, G. P. *Tetrahedron Lett.* **2001**, 42, 8939–8941. Bashiardes, G.; Safir, I.; Mohamed, A. S.; Barbot, F.; Laduranty, J. *Org. Lett.* **2003**, 5, 4915–4918.

<sup>(16)</sup> Broadly similar yields and *exo/endo* ratios were observed when these reactions were carried out (with and without ionic liquid) on a 1 mmol scale under "open vessel" conditions (PhMe, 200 W, 4 h).

<sup>(17)</sup> See ref 4a and: Grabowski, E. J. J.; Reider, P. J. Eur. Pat. 78026; *Chem. Abstr.* **1983**, *99*, 122171. Oxazolidinone **13** is not readily amenable to purification. The yield of **14c** observed then reflects the preparation of **13** as well as the cycloaddition step. We have noted<sup>4a</sup> that the presence of the C(6)-substituent does reduce the efficiency of the cycloaddition reactions as compared to reactions involving **3**.

<sup>(18)</sup> Attempts to improve the yield of cycloadducts derived from 13 were investigated. For example, cycloadduct 14a was generated in a one-pot process. This involved carrying out the initial diazo insertion step under microwave-assisted conditions, which were significantly faster (10 min vs 5 h) than without microwave irradiation, and 8a was added directly to crude 13 and subjected to our standard microwave cycloaddition conditions. This provided 14a but in a low (17%) yield.